# Enantioselective synthesis of 2-substituted 3-aminopropanoic acid ( $\beta$-alanine) derivatives which are $\beta$-analogues of aromatic amino acids 

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3-Aminopropanoic acid derivatives with a phenyl, 4-hydroxyphenyl, benzyl or indol-3-yl substituent at C-2 can be prepared enantioselectively by routes involving electrophilic attack of synthetic equivalents of $\left[\mathrm{H}_{2} \mathrm{NCH}_{2}\right]^{+}$upon enolates derived from chiral 3-acyl-1,3-oxazolidin-2-ones. tert-Butyl bromoacetate may be used as the electrophile, with subsequent introduction of nitrogen through the Curtius reaction, using the sequence of reagents (i) $\mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{H}$; (ii) $(\mathrm{PhO})_{2} \mathrm{P}(\mathrm{O}) \mathrm{N}_{3}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{PhCH}_{2} \mathrm{OH}$; alternatively, direct electrophilic reaction with 1-[ $N$-(benzyloxycarbonyl)aminomethyl]benzotriazole 4 c or benzyl $N$-(acetoxymethyl)carbamate 2d introduces a protected aminomethyl group in a single step.

## Introduction

There is considerable interest in the enantioselective synthesis of $\beta$-amino acids, compounds which can have biological activity in their own right and which are useful building blocks for the preparation of modified peptides and other potential pharmaceuticals. ${ }^{1}$ However, there have been few reports of asymmetric routes to $\beta$-alanine derivatives of general structure $\mathbf{1}$, substituted only at the $\alpha$-carbon. The known methods include

$1 \quad$ 2a $\mathrm{R}=\mathrm{PhCO} ; \mathrm{X}=\mathrm{Cl}$
$1 \quad$ 2b R $=\mathrm{PhCH}_{2} \mathrm{OCO}=\mathrm{Z} ; \mathrm{X}=\mathrm{OH}$
2c $\mathrm{R}=\mathrm{Z} ; \mathrm{X}=\mathrm{Cl}$
2d $\mathrm{R}=\mathrm{Z} ; \mathrm{X}=\mathrm{OAc}$


3a $\mathrm{R}=\mathrm{H}$
3b $\mathrm{R}=\mathrm{CH}_{2} \mathrm{NHCOPh}$

$4 \mathrm{a} \mathrm{R}=\mathrm{PhCO}$
4 b $\mathrm{R}=\mathrm{PhCH}_{2}$
4c $\mathrm{R}=\mathrm{Z}$
stereoselective alkylations of chiral enolates that are synthetically equivalent to the enolate from $\beta$-alanine ${ }^{2-4}$ and the conjugate addition of a carbon nucleophile to an $\alpha$-methylene $\beta$-alanine derivative. ${ }^{5}$ However, such approaches are unsuitable for the preparation of the $\beta$-amino acids $\mathbf{1}$ where R is an aryl group. In 1995 we described, in preliminary form, the first enantioselective synthesis of 3 -amino-2-phenylpropanoic acid ( $\alpha$-phenyl- $\beta$-alanine, 13). ${ }^{6}$ Our key step was the asymmetric Mannich reaction of the benzotriazole derivative $4 \mathbf{c}$ with the lithium enolate of the acyl oxazolidinone 6a. Recently Williams and co-workers reported that $\alpha$-aryl- $\beta$-alanines may also be prepared by sequences involving asymmetric Pd -catalysed reactions of allylic acetates with dimethyl malonate, ${ }^{7}$ which in this context are equivalent to the $\left[\mathrm{RCHCO}_{2} \mathrm{H}\right]^{+}$and $\left[\mathrm{H}_{2} \mathrm{NCH}_{2}\right]^{-}$ synthons respectively. Here we provide a full account of our complementary approaches to $\alpha$-substituted- $\beta$-alanines $\mathbf{1}$, in which chiral enolates undergo electrophilic attack by synthetic equivalents of the $\left[\mathrm{H}_{2} \mathrm{NCH}_{2}\right]^{+}$cation.

[^0]Evans et al. have reported the use of $N$-(chloromethyl)benzamide 2a to amidomethylate the titanium enolate of the acyl oxazolidinone 3a, leading to the stereoselective formation of the protected amino acid 3b in high yield. ${ }^{8}$ However, the reagent $\mathbf{2 a}$ is known to be somewhat unstable. ${ }^{9}$ Furthermore, the removal of the $N$-benzoyl group normally involves heating with mineral acid. These hydrolysis conditions could cause racemisation, particularly when enolisation is facilitated by the presence of an $\alpha$-aryl group, and they are likely to destroy sensitive aromatic groups such as the indolyl moiety.

1-(Aminomethyl)benzotriazoles and their N -acyl derivatives can also function as aminomethylating agents. ${ }^{10}$ Work by Page et al. ${ }^{11}$ has shown that the reagents $\mathbf{4 a}$ and $\mathbf{4 b}$, with $N$-benzoyl and $N$-benzyl protecting groups respectively, may be used to effect the stereoselective aminoalkylation of ketone enolates containing the 1,3 -dithiane-1-oxide auxiliary. 2-(Bromomethyl)phthalimide is commercially available and has been used for enolate alkylation, but again harsh conditions (e.g. refluxing $40 \% \mathrm{HBr})$ are needed for the $N$-deprotection. ${ }^{12}$ We favoured the $N$-benzyloxycarbonyl (Z) protecting group for use in the synthesis of the amino acids $\mathbf{1}$. The Z group survives the conditions for removal of common chiral auxiliaries, but may be cleaved by catalytic hydrogenolysis under mild conditions; because all the by-products of this deprotection are volatile, the final isolation of the amino acids is made extremely straightforward.

## Results and discussion

We elected to make use of the Evans oxazolidinone chiral auxiliary 5 to achieve stereoselectivity in attack upon the enolates (Scheme 1, Table 1). By analogy with simple alkylation, ${ }^{13}$ this was predicted to lead to the preferential formation of precursors to the $(R)-\beta$-amino acids 13-16; if the ( $S$ )-isomers of the amino acids were required, then they could be prepared using the enantiomer of 5 , which is readily available.
We decided to protect the phenolic hydroxy group of the tyrosine analogue $\mathbf{1 4}$ as a benzyl ether and so used the known ${ }^{14}$ [4-(benzyloxy)phenyl]acetyl chloride in the acylation step. We also considered it appropriate to protect the indolyl system in tryptophan analogue $\mathbf{1 6}$ against deprotonation and electrophilic attack by using a group which could be cleaved by hydrogenolysis: we chose to use the $N$-(indolyl) benzyloxycarbonyl group. This protecting group has occasionally been used in tryptophan chemistry, ${ }^{15}$ but the indole-3-acetic acid derivative $\mathbf{1 8}$ has not been described before and little has been said about the

Table 1 Percentage yields obtained in transformations depicted in Scheme 1 and diastereoselectivities (ds) observed in Mannich reactions.

|  | Yields (\%) |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $5 \longrightarrow 6$ | $6 \longrightarrow 8$ | $8 \longrightarrow 10$ | $6 \longrightarrow 10^{a}$ | $6 \longrightarrow 10^{\text {b }}$ | $10 \longrightarrow 12$ | $12 \longrightarrow(13-15)$ | Ds 10:11 from Mannich reaction |
| a | 80 | 67 | 58 | 65 | 62 | 65 | $68^{d}$ | 96:4 ${ }^{\text {a,e }}$ |
| b | 55 | 56 | 40 | 59 | - | 69 | 91 | 95:5 $5^{\text {a,e }}$ |
| c | 48 | 49 | 52 | 58 | - | 84 | - | 93:7; ${ }^{\text {a,e } 95: 5^{a, f}}$ |
| d | 83 | 69 | 47 | $<10^{\text {f }}$ | 38 | 67 | 81 | $>90: 10^{\text {b,f }}$ |
| e | 84 | - | - | ca. 0 | 34 | - | - |  |

${ }^{a}$ Using 4c. ${ }^{b}$ Using 2d. ${ }^{c}$ Yield based on 10a. ${ }^{d}$ Based on isolated yields of $\mathbf{1 0}$ and $\mathbf{1 1}$. ${ }^{e}$ Based on $250 \mathrm{MHz}{ }^{\mathbf{1}} \mathrm{H}$ NMR spectrum of crude reaction mixture.


Scheme 1 Reagents and conditions: i, BuLi, THF, $-78{ }^{\circ} \mathrm{C}$; ii, $\mathrm{RCH}_{2} \mathrm{COCl}$; iii, $\mathrm{NaN}\left(\mathrm{SiMe}_{3}\right)_{2}, \mathrm{THF},-78{ }^{\circ} \mathrm{C}$, then $\mathrm{BrCH}_{2} \mathrm{CO}_{2} \mathrm{Bu}^{t}$; iv, $\mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{H}, 20^{\circ} \mathrm{C}$, $1 \mathrm{~h} ; \mathrm{v},(\mathrm{PhO})_{2} \mathrm{P}(\mathrm{O}) \mathrm{N}_{3}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{PhCH}_{2} \mathrm{OH}, \mathrm{PhMe}, 20^{\circ} \mathrm{C}$, then reflux; vi, LDA or $\mathrm{LiN}\left(\mathrm{SiMe}_{3}\right)_{2}, \mathrm{THF},-78{ }^{\circ} \mathrm{C}$, then $\mathbf{4 c}$; vii, $\mathrm{TiCl}_{4}, \mathrm{EtNPr}_{2}{ }^{\circ}, \mathrm{ZNHCH}_{2} \mathrm{OAc}$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 20^{\circ} \mathrm{C}$; viii, $\mathrm{LiOH}, \mathrm{H}_{2} \mathrm{O}_{2}, \mathrm{H}_{2} \mathrm{O}, 0^{\circ} \mathrm{C}$; ix, $\mathrm{H}_{2}, \mathrm{Pd}-\mathrm{C}, \mathrm{AcOH}$
compatibility of 1-(benzyloxycarbonyl)indole derivatives with strongly basic and nucleophilic reagents. The dilithium salt of indole-3-acetic acid was acylated with benzyl chloroformate, thus introducing a Z group onto the indole nitrogen (Scheme 2).


Scheme 2 Reagents and conditions: i, $\operatorname{BuLi}$ (2.2 equiv.), THF, $-45^{\circ} \mathrm{C}$, then $\mathrm{PhCH}_{2} \mathrm{OCOCl}$ (1 equiv.); ii, $\mathrm{SOCl}_{2}, 20^{\circ} \mathrm{C}, 14 \mathrm{~h}$

The protected acid 18 was then converted into the acyl chloride 19, prior to coupling with the chiral auxiliary 5 in the usual way.

It has been reported that sodium and lithium enolates bearing chiral auxiliaries such as 5 undergo alkylation by $\alpha$-haloacetate esters with diastereoselectivity (ds) values usually $\geqslant 95: 5$ (ref. 16). We used this general method to convert the acyl oxazolidinones 6 into the tert-butyl esters 8 , all of which were highly crystalline compounds that could easily be obtained in isomerically pure form. We did not isolate the minor diastereoisomers 7, but in each case the 250 MHz NMR spectrum of the crude reaction mixture was studied and it was concluded that the amount of the by-product 7 could not have exceeded $10 \%$ of the amount of the major product $\mathbf{8}$. Treatment of the esters $\mathbf{8}$ with trifluoroacetic acid led to smooth deprotection to form the corresponding carboxylic acids 9 , which were transformed into the urethanes $\mathbf{1 0}$, using diphenylphosphoryl azide, triethylamine and benzyl alcohol in the general procedure of Shioiri et al. ${ }^{17}$ In this variant of the Curtius rearrangement the intermediate acyl azides and isocyanates need not be isolated. Com-
parison of the $250 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectrum of crude 10a from the Curtius reaction with an authentic sample of the diastereoisomer ent-11a (formed by the Mannich reaction as described below) demonstrated that no detectable epimerisation had occurred during the former reaction. Having obtained reference samples of the $N$-(benzyloxycarbonyl)- $\beta$-amino acid derivatives 10, we then sought methods for preparing these compounds directly from the acyloxazolidinones 6 .

The Mannich reagent $4 c$ was found to be stable, crystalline and easily prepared by refluxing commercially available 1 (hydroxymethyl)benzotriazole and benzyl carbamate together in toluene with a catalytic amount of toluene-4-sulfonic acid, using a Dean and Stark trap to remove water. The benzotriazolide anion is a relatively poor leaving group ( $\mathrm{p} K_{\mathrm{a}}$ of benzotriazole $=8.2)^{18}$ and 4 c does not show high reactivity as an electrophile. Thus $\mathbf{4 c}$ completely failed to react with the titanium enolate of $\mathbf{6 a}$ at $20^{\circ} \mathrm{C}$. Reaction of $\mathbf{4 c}$ with the Li and Na enolates of $\mathbf{6 a}$, which were quenched at temperatures of -10 and $-5^{\circ} \mathrm{C}$ respectively, gave Mannich product 10a in yields of only 31 and $45 \%$; again much of the acyl oxazolidinone 6a remained unchanged. The highest yield ( $65 \%$ ) of the desired product 10a was obtained when a mixture containing $\mathbf{4 c}$ and the lithium enolate of $\mathbf{6 a}$ was allowed to warm up to $25^{\circ} \mathrm{C}$ before being quenched; under these conditions the major oxazolidinone-containing by-product was the deacylated chiral auxiliary 5 , which can arise by the known decomposition of enolates at higher temperatures through a ketene mechanism. ${ }^{13}$ Analysis of the crude product by $250 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy indicated that 10a, $\mathbf{5}$ and unreacted $\mathbf{6 a}$ were present in the approximate molar ratio $8: 2: 1$. Very little of the diastereoisomeric Mannich product 11a was formed and the similarity in the chemical shifts of 11a to those of the main products made it difficult to detect 11a in the NMR spectrum of the crude product. When we performed the Mannich reaction using ent-6a we were able to isolate the minor product ent-11a in $2 \%$ yield. Reagent $\mathbf{4 c}$ was also successfully employed to effect the direct conversion of the lithium enolates of $\mathbf{6 b}$ and $\mathbf{6 c}$ into the urethanes 10b and 10c (yields 59 and $58 \%$ respectively). Again only small amounts (ca. $3 \%$ ) of the minor diastereoisomers 11b and 11c were formed. Cleavage of the Z group from the indole nitrogen did not occur under these conditions. Attempted reactions of lithium enolates from the (3-phenylpropanoyl)oxazolidinone $\mathbf{6 d}$ and the propanoyloxazolidinone 6 e with the benzotriazole derivative $\mathbf{4 c}$ did not yield synthetically useful amounts of Mannich products 10 and 11: the acyl oxazolidinones $\mathbf{6}$ were mainly recovered unchanged, although some of the chiral auxiliary 5 was observed as a minor product. Thus the Mannich reaction was successful in cases where R was an aryl group, but not when R was alkyl. This suggested that those lithium enolates which are not stabilised by conjugation with an $\alpha$-aryl group may be sufficiently basic to abstract the $\mathrm{N}-\mathrm{H}$ proton from the benzotriazole reagent $\mathbf{4 c}$. We therefore examined the possibility of introducing the $\mathrm{ZNHCH}_{2}$ group in Lewis acidpromoted Mannich reactions.

The benzotriazole-derived reagent $4 \mathbf{c}$ was found to be unreactive towards the titanium enolate of $\mathbf{6 a}$. Benzyl N -(hydroxymethyl)carbamate $\left(\mathrm{ZNHCH}_{2} \mathrm{OH}\right) \mathbf{2 b}$ was easily prepared, ${ }^{19}$ but our attempts to convert it into the chloride $\mathrm{ZNHCH}_{2} \mathrm{Cl} 2 \mathbf{2 c}$, using $\mathrm{PCl}_{5}$, by analogy with the preparation of $\mathbf{2 a}$, led only to decomposition. However, the acetate 2d is readily available and comparatively stable. We found that this reagent was able to bring about the stereoselective Mannich reactions on titanium enolates both when $R$ was aryl $(\mathbf{6 a} \longrightarrow \mathbf{1 0 a}, 62 \%)$ and when $R$ was alkyl. In the latter cases the yields were rather low ( $\mathbf{6 d} \longrightarrow \mathbf{1 0 d}, 38 \% ; \mathbf{6 e} \longrightarrow \mathbf{1 0 e}, 34 \%$ ) and much of the starting acyl oxazolidinone remained unconverted, even when the reaction mixture was allowed to warm up to room temperature (e.g $42 \%$ of the starting acyl oxazolidinone $\mathbf{6 d}$ was recovered after 1 h at $20^{\circ} \mathrm{C}$ ).
Removal of the chiral auxiliaries from compounds 10a-d was
performed under Evans' usual hydrolytic conditions ( LiOH , $\mathrm{H}_{2} \mathrm{O}_{2}, \mathrm{H}_{2} \mathrm{O}$ ) and gave the carboxylic acids 12a-d in yields of $\geqslant 65 \%$ after recrystallisation; if mild conditions were used for the hydrolysis then the Z group on the indole system was not affected. Finally the free amino acids $\mathbf{1 3}, \mathbf{1 4}$ and $\mathbf{1 5}$ were prepared by catalytic hydrogenolysis.
The specific rotation of the ( $R$ )-3-amino-2-phenylpropanoic acid $\mathbf{1 3}$ was equal in both sign and magnitude of that of resolved material which had been assigned the ( $S$ )-absolute configuration on the basis of a lengthy series of chemical correlations. ${ }^{20}$ However, a single crystal X-ray diffraction study ${ }^{6}$ of the salt of $\mathbf{1 3}$ with ( $1 S$ )-(+)-camphor-10-sulfonic acid showed that we had indeed synthesised $(R)-\mathbf{1 3}$ and that the earlier assignment of absolute configuration by Garbarino and Nuñez was incorrect.
In our hands $(R)-\mathbf{1 5}$ had $[a]_{\mathrm{D}}+19$ in 1 m HCl . This is somewhat larger than the literature value of +11.3 , which has been reported by Juaristi et al. ${ }^{2}$ Juaristi has observed that the specific rotations of free $\beta$-amino acids are rather sensitive to the conditions of measurement. ${ }^{21}$ We wondered if it might be possible that partial racemisation occurred in the final step of Juaristi's synthesis (hydrolysis by $6 \mathrm{~m} \mathrm{HCl}, 90-100^{\circ} \mathrm{C}, 8 \mathrm{~h}$ ). When we heated a sample of $\mathbf{1 5}$ in 6 m DCl in an NMR tube at $93-97^{\circ} \mathrm{C}$ for 7 days and studied the 600 MHz NMR spectrum, we found that no decomposition of $\mathbf{1 5}$ had occurred, but that the changes in integration and appearance of the peaks in the region $\delta_{\mathrm{H}} 2.5-$ 3.2 were consistent with ca. $35 \%$ deuterium incorporation at the chiral centre. Thus there is a real possibility that small amounts of racemisation may occur when $\alpha$-substituted $\beta$-alanine derivatives are exposed to hot hydrochloric acid.

## Conclusions

We have established synthetically useful routes to enantiomerically pure, $\alpha$-substituted $\beta$-alanine derivatives. These methods are especially attractive because they deliver the amino acids in Z-protected form, convenient for further synthesis or for deprotection under mild conditions by hydrogenolysis. They avoid the use of refluxing mineral acid for deprotection, thus minimising the risk of racemisation.
In cases where the $\alpha$-substituent is an aryl group, the protected aminomethyl group is conveniently introduced by reaction of an appropriate chiral lithium enolate with the benzotriazole-derived Mannich reagent $\mathbf{4 c}$. This direct approach fails when the $\alpha$-substituent is an alkyl group, but under such circumstances the alternative Mannich reagent 2d, with acetate as the leaving group, can be used in conjunction with a titanium enolate. There may be scope for increasing the conversions obtained in these titanium-mediated reactions by further changing the leaving group.
$\beta$-Alanine derivatives with either alkyl or aryl substituents at the $\alpha$-position are also available by reactions of chiral enolates with tert-butyl bromoacetate, which can be rendered synthetically equivalent to a Mannich electrophile by a sequence of steps including a Curtius rearrangement. This method is less direct than the Mannich approach, but it tolerates a range of substituents (both alkyl and aryl) and the major diastereoisomer from the asymmetric step tends to be amenable to purification by crystallisation.

## Experimental

General experimental procedures have been described by us in an earlier publication. ${ }^{22}[a]_{\mathrm{D}}$ Values are given in units of $10^{-1}$ deg $\mathrm{cm}^{2} \mathrm{~g}^{-1}$. Hydrogenations were performed using balloons. All reported compounds were homogeneous as judged by both TLC and NMR spectroscopy. Mass spectra were obtained by electron impact unless otherwise stated. The known acyl oxazolidinones 6a (ref. 23), 6d (ref. 24) and 6e (ref. 25) were prepared from the appropriate acyl chlorides and the lithium
salt of $\mathbf{5}$, according to the general procedure of Evans et al. ${ }^{25}$ Petrol refers to light petroleum with bp $40-60^{\circ} \mathrm{C}$.

## 1-[ $N$-(Benzyloxycarbonyl)aminomethyl]benzotriazole 4c

1-(Hydroxymethyl)benzotriazole ( $3.89 \mathrm{~g}, 26.1 \mathrm{mmol}$ ), benzyl carbamate ( $3.94 \mathrm{~g}, 26.1 \mathrm{mmol}$ ), toluene-p-sulfonic acid monohydrate ( 0.01 g ) and toluene ( 70 ml ) were refluxed together for 15 h in an apparatus fitted with a Dean and Stark water separator. The mixture was cooled and the crystals which separated were filtered off. Recrystallisation from toluene gave the title compound $4 \mathrm{c}(4.53 \mathrm{~g}, 62 \%)$ as a white crystalline solid, $\mathrm{mp} 119-$ $120^{\circ} \mathrm{C}$ (Found: C, 64.0; H, 5.0; N, 20.0. $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires C, 63.8; H, 5.0; N, 19.85\%); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 3255,1725,1531$ and $1251 ; \delta_{\mathrm{H}}\left(80 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 5.15(2 \mathrm{H}, \mathrm{s}), 6.05(2 \mathrm{H}, \mathrm{d}, J 7), 6.5(1$ $\mathrm{H}, \mathrm{br} \mathrm{t}, J 7), 7.2-7.6(7 \mathrm{H}, \mathrm{m})$ and $7.8-8.1(2 \mathrm{H}, \mathrm{m}) ; m / z 282$ $\left(\mathrm{M}^{+}, 15 \%\right), 119$ (54) and 91 (100) (Found: $\mathrm{M}^{+}, 282.1115$. $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M$, 282.1117).

## (4S,5R)-3-[4-(Benzyloxy)phenylacetyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 6b

A solution of $(4 S, 5 R)$-4-methyl-5-phenyl-1,3-oxazolidin-2-one $(5.10 \mathrm{~g}, 28.8 \mathrm{mmol})$ in THF ( 30 ml ) was cooled to $-78^{\circ} \mathrm{C}$ and treated over 2 min with 2.5 m BuLi in hexane $(12.7 \mathrm{ml}$, 31.7 mmol ), followed by a solution of 4-(benzyloxy)phenylacetyl chloride ${ }^{14}(7.52 \mathrm{~g}, 28.8 \mathrm{mmol})$ in THF ( 30 ml ), which was added during 2 min . The reaction mixture was allowed to warm to $0^{\circ} \mathrm{C}$ over 1 h , then was quenched with saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}(5 \mathrm{ml})$ and partitioned between $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{ml})$ and water ( 100 ml ). The organic phase was washed with aqueous $\mathrm{NaHCO}_{3}(2 \times 100 \mathrm{ml})$, dried $\left(\mathrm{MgSO}_{4}\right)$, and evaporated to leave an orange oil. Flash chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\right.$ petrol (1:1) to $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{Et}_{2} \mathrm{O}$ (9:1); gradient elution] followed by recrystallisation from $\mathrm{Et}_{2} \mathrm{O}$-petrol gave (4S,5R)-3-[4-(benzoyloxy)phenyl-acety][-4-methyl-5-phenyl-1,3-oxazolidin-2-one $\mathbf{6 b}$ ( $6.36 \mathrm{~g}, 55 \%$ ) as white crystals, $\mathrm{mp} 97-98^{\circ} \mathrm{C}$ (Found: C, 74.8 ; H, 5.7; N, 3.4. $\mathrm{C}_{25} \mathrm{H}_{23} \mathrm{NO}_{4}$ requires C, $\left.74.8 ; \mathrm{H}, 5.7 ; \mathrm{N}, 3.5 \%\right) ;[a]_{\mathrm{D}}^{30}+6.3(c 1.0$, $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 1778$ and $1698 ; \delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $0.88(3 \mathrm{H}, \mathrm{d}, J 7), 4.21(1 \mathrm{H}, \mathrm{d}, J 15), 4.28(1 \mathrm{H}, \mathrm{d}, J 15), 4.75(1$ H, quintet, $J$ 7), $5.05(2 \mathrm{H}, \mathrm{s}), 5.64(1 \mathrm{H}, \mathrm{d}, J 7), 6.92-6.98(2 \mathrm{H}$, $\mathrm{m})$ and $7.22-7.45(14 \mathrm{H}, \mathrm{m}) ; m / z 401\left(\mathrm{M}^{+}, 4 \%\right), 224(55)$ and 91 (100) (Found: $\mathrm{M}^{+}, 401.1638 . \mathrm{C}_{25} \mathrm{H}_{23} \mathrm{NO}_{4}$ requires $M, 401.1627$ ).

## (4S,5R)-3-[(1-Benzyloxycarbonylindol-3-yl)acetyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 6 c

[1-(Benzyloxycarbonyl)indol-3-yl]acetic acid $18(1.17 \mathrm{~g}, 3.79$ $\mathrm{mmol})$ was dissolved in $\mathrm{SOCl}_{2}(11 \mathrm{ml})$ and the mixture was allowed to stir overnight. The excess of $\mathrm{SOCl}_{2}$ and the side products were evaporated under vacuum to leave a dark coloured oil ( 1.27 g ), considered to be [1-(benzyloxycarbonyl)-indol-3-yl]acetyl chloride $\mathbf{1 9}$ on the basis of the following data: $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 1798$ and $1737 ; \delta_{\mathrm{H}}\left(60 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 4.2(2 \mathrm{H}, \mathrm{s})$, $5.4(2 \mathrm{H}, \mathrm{s}), 7.2-7.5(8 \mathrm{H}, \mathrm{m}), 7.6(1 \mathrm{H}, \mathrm{s})$ and $8.1-8.3(1 \mathrm{H}, \mathrm{m})$. A portion of crude compound $19(1.24 \mathrm{~g}, 3.79 \mathrm{mmol})$ was used to acylate the oxazolidinone 5 by analogy with the preparation of 6b. Flash chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\right.$ petrol (7:3)] gave the title compound $\mathbf{6 c}(0.854 \mathrm{~g}, 48 \%)$ as a white solid, $\mathrm{mp} 52-54^{\circ} \mathrm{C} ;[a]_{\mathrm{D}}^{35}$ -0.4 (c 1.0, $\left.\mathrm{CHCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 1778,1732$ and 1705 ; $\delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 0.90(3 \mathrm{H}, \mathrm{d}, J 7), 4.36(1 \mathrm{H}, \mathrm{dd}, J 17,1)$, $4.44(1 \mathrm{H}, \mathrm{dd}, J 17,1), 4.77(1 \mathrm{H}$, quintet, $J 7), 5.45(2 \mathrm{H}, \mathrm{s}), 5.67$ $(1 \mathrm{H}, \mathrm{d}, J 7), 7.24-7.64(13 \mathrm{H}, \mathrm{m}), 7.70(1 \mathrm{H}, \mathrm{s})$ and $8.20(1 \mathrm{H}, \mathrm{d}$, $J 8) ; m / z 468$ ( $\left.\mathrm{M}^{+}, 19 \%\right), 291$ (6), 247 (27), 220 (7), 157 (11), 91 (100), 65 (6) and 44 (11) (Found: $\mathrm{M}^{+}, 468.1683 . \mathrm{C}_{28} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{5}$ requires $M, 468.1685$ ).

## (4S,5R,2'S)-3-(3-tert-Butoxycarbonyl-2-phenylpropanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 8a

A 1 m solution of $\mathrm{NaN}\left(\mathrm{SiMe}_{3}\right)_{2}$ in THF (Aldrich; 10.2 ml , 10.2 mmol ) was diluted with dry THF ( 10 ml ) and cooled to $-72^{\circ} \mathrm{C}$. A solution of ( $4 S, 5 R$ )-4-methyl-5-phenyl-3-phenylacetyl-1,3-oxazolidin-2-one $\mathbf{6 a}{ }^{23}(2.50 \mathrm{~g}, 8.5 \mathrm{mmol})$ in THF ( 20 ml ) was
then added by cannula over 5 min . The mixture was stirred during 30 min and then tert-butyl bromoacetate ( $1.65 \mathrm{ml}, 10.2$ mmol ) was added by syringe. The mixture was allowed to attain $-20^{\circ} \mathrm{C}$ over 2 h and was then quenched with saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}(5 \mathrm{ml})$.
The mixture was extracted with ethyl acetate ( 100 ml ), washed with distilled water $(100 \mathrm{ml})$, and the organic layer was dried. The solvent was evaporated under vacuum to yield an orange, viscous liquid, which was purified by flash chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$-petrol ( $80: 20$ ) to ( $90: 10$ ); gradient elution] and crystallisation (EtOAc-petrol) to yield the title compound $8 \mathbf{a}(2.18 \mathrm{~g}, 67 \%)$ as white crystals, $\mathrm{mp} 125^{\circ} \mathrm{C} ;[a]_{\mathrm{D}}^{30}+63.8(c 0.5$, $\left.\mathrm{CHCl}_{3}\right) ; v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1767,1722$ and $1700 ; \delta_{\mathrm{H}}(80 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 0.97(3 \mathrm{H}, \mathrm{d}, J 6), 1.45(9 \mathrm{H}, \mathrm{s}), 2.60(1 \mathrm{H}, \mathrm{dd}, J 17,4.5)$, $3.28(1 \mathrm{H}, \mathrm{dd}, J 17,11), 4.70(1 \mathrm{H}$, quintet, $J 7), 5.4-5.7(2 \mathrm{H}, \mathrm{m})$ and 7.2-7.5 (10 H, m); m/z $409\left(\mathrm{M}^{+}, 10 \%\right), 336$ (10), 292 (10), 178 (100), 134 (11), 104 (10) and 57 (40) (Found: $\mathrm{M}^{+}, 409.1894$. $\mathrm{C}_{24} \mathrm{H}_{27} \mathrm{NO}_{5}$ requires $M, 409.1889$ ).

## (4S,5R,2'S)-3-[2-(4-Benzyloxyphenyl)-3-(tert-butoxycarbonyl)-

 propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 8bThis was prepared by analogy with 8a, starting from $(4 S, 5 R)$-3-[4-(benzyloxy)phenylacetyl]-4-methyl-5-phenyl-1,3-oxazolidin2 -one $\mathbf{6 b}(1.60 \mathrm{~g}, 4.0 \mathrm{mmol})$. The title compound $\mathbf{8 b}(1.18 \mathrm{~g}, 56 \%)$ was obtained as white crystals, mp $158-160^{\circ} \mathrm{C}$ (from EtOAcpetrol); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 1783,1732$ and $1701 ; \delta_{\mathrm{H}}(250 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 0.93(3 \mathrm{H}, \mathrm{d}, J 7), 1.40(9 \mathrm{H}, \mathrm{s}), 2.56(1 \mathrm{H}, \mathrm{dd}, J 17.5,5)$, $3.21(1 \mathrm{H}, \mathrm{dd}, J 17.5,11), 4.67(1 \mathrm{H}$, quintet, $J$ 7), $5.40(2 \mathrm{H}, \mathrm{s})$, $5.45(1 \mathrm{H}, \mathrm{dd}, J 11,5), 5.50(1 \mathrm{H}, \mathrm{d}, J 7), 6.90-6.97(2 \mathrm{H}, \mathrm{m})$ and $7.25-7.45(12 \mathrm{H}, \mathrm{m}) ; m / z 515\left(\mathrm{M}^{+}, 1 \%\right), 282(21), 178$ (15) and 91 (100) (Found: $\mathrm{M}^{+}$, 515.2314. $\mathrm{C}_{31} \mathrm{H}_{33} \mathrm{NO}_{6}$ requires $M$, 515.2308).

## (4S,5R,2'S)-3-[2-(1-Benzyloxycarbonylindol-3-yl)-3-(tert-butoxycarbonyl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2one 8 c

This was prepared by analogy with $\mathbf{8 a}$, starting from $(4 S, 5 R)-3-$ [(1-benzyloxycarbonylindol-3-yl)acetyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one $6 \mathbf{c}(0.72 \mathrm{~g}, 1.53 \mathrm{mmol})$. Purification of the crude product by flash chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$-petrol (4:1)] and recrystallisation $\left(\mathrm{Et}_{2} \mathrm{O}\right.$-petrol) yielded the title compound $8 \mathrm{c}(0.439 \mathrm{~g}, 49 \%)$ as white crystals, $\mathrm{mp} 145^{\circ} \mathrm{C}$ (Found: C, 70.1 ; $\mathrm{H}, 5.9 ; \mathrm{N}, 4.8 . \mathrm{C}_{34} \mathrm{H}_{34} \mathrm{~N}_{2} \mathrm{O}_{7}$ requires C, 70.1; H, 5.9; N, 4.8\%); $[a]_{\mathrm{D}}^{30}+78.5\left(c 1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 1765,1734$ and 1705 ; $\delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 0.95(3 \mathrm{H}, \mathrm{d}, J 7), 1.41(9 \mathrm{H}, \mathrm{s}), 2.65(1 \mathrm{H}$, dd, $J 17,5), 3.40(1 \mathrm{H}, \mathrm{dd}, J 17,11), 4.69(1 \mathrm{H}$, quintet, $J 7$ ), $5.38-5.50(3 \mathrm{H}, \mathrm{m}), 5.74(1 \mathrm{H}, \mathrm{dd}, J 11,5), 7.25-7.50(12 \mathrm{H}, \mathrm{m})$, $7.66(1 \mathrm{H}, \mathrm{s}), 7.80(1 \mathrm{H}, \mathrm{d}, J 8)$ and $8.20(1 \mathrm{H}, \mathrm{d}, J 8) ; m / z 582$ ( $\mathrm{M}^{+}, 0.5 \%$ ), 526 (3), 349 (7), 278 (3), 233 (5) and 91 (100).
(4S,5R,2'R)-3-[2-Benzyl-3-(tert-butoxycarbonyl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 8d
This was prepared by analogy with 8a, starting from $(4 S, 5 R)$ -4-methyl-5-phenyl-3-(3-phenylpropanoyl)-1,3-oxazolidin-2-one $\mathbf{6 d}^{24}(2.50 \mathrm{~g}, 8.1 \mathrm{mmol})$. Recrystallisation from $\mathrm{Et}_{2} \mathrm{O}$-petrol yielded the title compound $\mathbf{8 d}$ as white crystals ( $2.35 \mathrm{~g}, 69 \%$ ), mp $79^{\circ} \mathrm{C} ;[a]_{\mathrm{D}}^{30}+45\left(c 0.2, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$; $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1766$ and 1706 ; $\delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 0.87(3 \mathrm{H}, \mathrm{d}, J 7), 1.38(9 \mathrm{H}, \mathrm{s}), 2.37(1 \mathrm{H}$, dd, $J 17,5), 2.71(1 \mathrm{H}, \mathrm{dd}, J 13,8), 2.82(1 \mathrm{H}, \mathrm{dd}, J 17,11), 2.99$ $(1 \mathrm{H}, \mathrm{dd}, J 13,7), 4.48-4.59(1 \mathrm{H}, \mathrm{m}), 4.60(1 \mathrm{H}$, quintet, $J 7$ ), $5.30(1 \mathrm{H}, \mathrm{d}, J 7)$ and $7.20-7.45(10 \mathrm{H}, \mathrm{m}) ; \mathrm{m} / \mathrm{z} 423\left(\mathrm{M}^{+}, 3 \%\right)$, 367 (94), 308 (86), 178 (68), 117 (72) and 57 (100) (Found: $\mathrm{M}^{+}$, 423.2040. $\mathrm{C}_{25} \mathrm{H}_{29} \mathrm{NO}_{5}$ requires $M, 423.2046$ ).

## (4S,5R,2'S)-3-(3-Carboxy-2-phenylpropanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 9a

( $4 S, 5 R, 2^{\prime} S$ )-3-(3-tert-Butoxycarbonyl-2-phenylpropanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 8a ( $103 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) was dissolved in trifluoroacetic acid $(1 \mathrm{ml})$ at room temperature. After 1 h , the pale pink solution was evaporated under vacuum.

The residue was dissolved in diethyl ether and was then reevaporated. Recrystallisation from $\mathrm{Et}_{2} \mathrm{O}$-petrol yielded the title compound 9a ( $101 \mathrm{mg}, 98 \%$ ) as white crystals, $\mathrm{mp} 151^{\circ} \mathrm{C}$; $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 2700-3500,1765,1734$ and $1694 ; \delta_{\mathrm{H}}(80 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 0.95(3 \mathrm{H}, \mathrm{d}, J 7), 2.7(1 \mathrm{H}, \mathrm{dd}, J 18,4), 3.45(1 \mathrm{H}, \mathrm{dd}$, $J 18,12), 4.75$ (1 H, quintet, $J$ 7), $5.4-5.7(2 \mathrm{H}, \mathrm{m}), 7.3-7.6$ (10 $\mathrm{H}, \mathrm{m})$ and $8.4(1 \mathrm{H}, \mathrm{br} \mathrm{s}) ; m / z 353\left(\mathrm{M}^{+}, 30 \%\right), 177(21), 148$ (6) and 107 (100) (Found: $\mathrm{M}^{+}, 353.1259 . \mathrm{C}_{20} \mathrm{H}_{19} \mathrm{NO}_{5}$ requires $M$, 353.1263).

## (4S,5R,2'R)-3-(2-Benzyl-3-carboxypropanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 9d

By analogy with the preparation of $9 \mathbf{9 a},\left(4 S, 5 R, 2^{\prime} R\right)$-3-[2-benzyl-3-(tert-butoxycarbonyl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one $8 \mathbf{d}$ ( $1.00 \mathrm{~g}, 2.36 \mathrm{mmol}$ ) was converted into the title compound 9d, which was obtained as a white foam ( $682 \mathrm{mg}, 79^{\%} \%$ ), $\mathrm{mp} 62^{\circ} \mathrm{C}$; $[a]_{\mathrm{D}}^{30}+52\left(c 0.2, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ ); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 2800-3600(\mathrm{br}), 1781$ and $1703 ; \delta_{\mathrm{H}}(250 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 0.84(3 \mathrm{H}, \mathrm{d}, J 7), 2.47(1 \mathrm{H}, \mathrm{dd}, J 17,5), 2.68(1 \mathrm{H}$, dd, $J 13,9$ ), $2.92(1 \mathrm{H}$, dd, $J 17,10), 3.03(1 \mathrm{H}, \mathrm{dd}, J 13,7)$, 4.45-4.59 (1 H, m), $4.63(1 \mathrm{H}$, quintet, $J 7), 5.36(1 \mathrm{H}, \mathrm{d}, J 7)$ and 7.20-7.45 (10 H, m); m/z $367\left(\mathrm{M}^{+}, 15 \%\right), 308$ (17), 177 (21), 148 (31) and 107 (100) (Found: $\mathrm{M}^{+}, 367.1420 . \mathrm{C}_{21} \mathrm{H}_{21}{ }^{-}$ $\mathrm{NO}_{5}$ requires $M, 367.1420$ ).

## (4S,5R,2'R)-3-(3-Benzyloxycarbonylamino-2-phenylpropanoyl)-

 4-methyl-5-phenyl-1,3-oxazolidin-2-one 10a(i) Preparation of 10 a via the Curtius reaction. $\left(4 S, 5 R, 2^{\prime} S\right)$-3-(3-Carboxy-2-phenylpropanoyl)-4-methyl-5-phenyl-1,3-oxazol-idin-2-one 9a ( $404 \mathrm{mg}, 1.14 \mathrm{mmol}$ ) was dissolved in toluene ( 5 $\mathrm{ml})$. Triethylamine ( $320 \mu \mathrm{l}, 2.3 \mathrm{mmol}$ ), diphenylphosphoryl azide ( $300 \mu \mathrm{l}, 1.4 \mathrm{mmol}$ ) and benzyl alcohol ( $241 \mu \mathrm{l}, 2.3 \mathrm{mmol}$ ) were added. The mixture was stirred for 1 h before being refluxed for 1 h . Evaporation of the solvent left a brown residue, which was dissolved in diethyl ether $(14 \mathrm{ml})$ and washed with 2 $\mathrm{m} \mathrm{HCl}(7 \mathrm{ml})$ and then with saturated aqueous $\mathrm{NaHCO}_{3}(7 \mathrm{ml})$. The organic layer was dried and then evaporated under vacuum to yield a brown oil, which was purified by repeated flash chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{Et}_{2} \mathrm{O}\right.$ (99:1)] to yield the title compound $10 \mathrm{a}(309 \mathrm{mg}, 59 \%)$ as a white foam, $\mathrm{mp} 50-51^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}^{29}$ +79 ( c 1, EtOAc). This material was identical by IR and 250 $\mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy to the major product 10 a obtained in the following experiment.
(ii) Preparation of 10 a via the Mannich reaction with $1-[N-$ (benzyloxycarbonyl)aminomethyl]benzotriazole 4c. A solution of (4S,5R)-4-methyl-5-phenyl-3-phenylacetyl-1,3-oxazolidin-2one $\mathbf{6 a}(295 \mathrm{mg}, 1.0 \mathrm{mmol})$ in THF $(4 \mathrm{ml})$ was added to LDA in THF ( $1.5 \mathrm{~m}, 0.73 \mathrm{ml}, 1.1 \mathrm{mmol}$ ) and the mixture was stirred for 20 min . A solution of 1-[ $N$-(benzyloxycarbonyl)aminomethyl]benzotriazole $\mathbf{4 c}(311 \mathrm{mg}, 1.1 \mathrm{mmol})$ in THF $(4 \mathrm{ml})$ was added and the mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for 2 h , then allowed to warm to $20^{\circ} \mathrm{C}$. Saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{ml})$ was added and the product was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{ml})$. The $\mathrm{Et}_{2} \mathrm{O}$ extracts were washed with $2 \mathrm{~m} \mathrm{HCl}(20 \mathrm{ml})$, aqueous $\mathrm{NaHCO}_{3}$ $(2 \times 20 \mathrm{ml})$ and brine $(20 \mathrm{ml})$. Drying and evaporation of the combined $\mathrm{Et}_{2} \mathrm{O}$ extracts gave a crude product whose ${ }^{1} \mathrm{H}$ NMR spectrum $\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ was consistent with the presence of $\quad\left(4 S, 5 R, 2^{\prime} R\right)$-3-(3-benzyloxycarbonylamino-2-phenylprop-anoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 10a, ( $4 S, 5 R$ )-4-methyl-5-phenyl-1,3-oxazolidin-2-one 5 and the starting material 6a as the three main components in an approximate molar ratio of $8: 2: 1$.

Flash chromatography [EtOAc-petrol $(1: 4)$ to $(2: 3)$; gradient elution] gave ( $4 \mathrm{~S}, 5 \mathrm{R}, 2^{\prime} \mathrm{R}$ )-3-(3-benzyloxycarbonylamino-2-phenylpropanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 10a $(297 \mathrm{mg}, 65 \%)$ as a foam, $[a]_{\mathrm{D}}^{30}+84(c 1.1, \mathrm{EtOAc}) ; v_{\max }(\mathrm{KBr}) /$ $\mathrm{cm}^{-1} 3368,1782,1721$ and $1698 ; \delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 0.92$ (3 $\mathrm{H}, \mathrm{d}, J 7), 3.56-3.67(1 \mathrm{H}, \mathrm{m}), 3.74-3.85(1 \mathrm{H}, \mathrm{m}), 4.68(1 \mathrm{H}$, quintet, $J$ 7), $5.00-5.13(3 \mathrm{H}, \mathrm{m}), 5.22(1 \mathrm{H}, \mathrm{dd}, J 8,6), 5.46$ $(1 \mathrm{H}, \mathrm{d}, J 7)$ and $7.21-7.44(15 \mathrm{H}, \mathrm{m}) ; m / z 458\left(\mathrm{M}^{+}, 1 \%\right), 307$
(8), 295 (100), 177 (4) and 118 (25) (Found: $\mathrm{M}^{+}, 458.1854$. $\mathrm{C}_{27} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{5}$ requires $M, 458.1842$ ).

A similar procedure to the above was used to convert ent-6a $(414 \mathrm{mg}, 1.40 \mathrm{mmol})$ into ent-10a ( $347 \mathrm{mg}, 54 \%$ ). Also isolated were: unreacted ent-6a ( $27 \mathrm{mg}, 7 \%$ ), ent-5 ( $51 \mathrm{mg}, 20 \%$ ) and (4R,5S,2'R)-3-(3-benzyloxycarbonylamino-2-phenylpropanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one ent-11a (14 mg, $2 \%$ ), isolated as a colourless oil and purified by flash chromatography in $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{Et}_{2} \mathrm{O}$ (97.5:2.5); $[\alpha]_{\mathrm{D}}^{25}+96$ (c 0.49 , EtOAc); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 3359,1781,1721$ and $1696 ; \delta_{\mathrm{H}}(250 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 0.75(3 \mathrm{H}, \mathrm{d}, J 7), 3.58-3.80(2 \mathrm{H}, \mathrm{m}), 4.81(1 \mathrm{H}$, quintet, $J 7), 5.05-5.13(4 \mathrm{H}, \mathrm{m}), 5.62(1 \mathrm{H}, \mathrm{d}, J 7)$ and $7.15-7.40(15 \mathrm{H}$, $\mathrm{m}) ; \mathrm{m} / \mathrm{z} 458\left(\mathrm{M}^{+}, 1 \%\right), 414(0.3), 367$ (1), 295 (100) and 91 (54) (Found: $\mathrm{M}^{+}, 458.1843 . \mathrm{C}_{27} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{5}$ requires $M, 458.1842$ ).
(iii) Preparation of 10 a via the Mannich reaction with benzyl N -(acetoxymethyl)carbamate. $\mathrm{TiCl}_{4}\left(10 \% \mathrm{v} / \mathrm{v}\right.$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2} ; 0.97$ $\mathrm{ml}, 0.88 \mathrm{mmol})$ was added to a solution of $(4 S, 5 R)-4$-methyl-5-phenyl-3-phenylacetyl-1,3-oxazolidin-2-one 6a ( $240 \mathrm{mg}, 0.81$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{ml})$ at $0{ }^{\circ} \mathrm{C}$. The pale orange mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 10 min and then $\operatorname{EtNPr}_{2}^{\mathrm{i}}(167 \mu 1,0.96 \mathrm{mmol})$ was added. The purple solution was stirred at $0^{\circ} \mathrm{C}$ for 1 h before being cooled to $-78^{\circ} \mathrm{C}$ and treated with a solution prepared from $\mathrm{TiCl}_{4}\left(10 \% \mathrm{v} / \mathrm{v}\right.$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2} ; 1.16 \mathrm{ml}, 1.06 \mathrm{mmol}\right)$, benzyl $N$-(acetoxymethyl)carbamate ${ }^{19}$ ( $230 \mathrm{mg}, 0.96 \mathrm{mmol}$ ) and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{ml})$ which had been kept at $0^{\circ} \mathrm{C}$ for 30 min . The reaction mixture was allowed to warm from -78 to $20^{\circ} \mathrm{C}$ over 6 h and then was stirred at $20^{\circ} \mathrm{C}$ for 1 h , before being quenched with aqueous $\mathrm{NH}_{4} \mathrm{Cl}(10 \mathrm{ml})$. The product was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{ml})$. Drying and evaporation of the combined $\mathrm{Et}_{2} \mathrm{O}$ extracts gave a yellow oil, which was subjected to flash chromatography [EtOAc-petrol; gradient from $1: 9$ to $3: 7$ ] to yield (4S,5R,2'R)-3-(3-benzyloxycarbonylamino-2-phenylprop-anoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 10a (230 mg, $62 \%$ ) as a pale yellow foam, identical by ${ }^{1} \mathrm{H}$ NMR spectroscopy ( $80 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) to the sample prepared in the preceeding experiment.

## (4S,5R,2'R)-3-[3-Benzyloxycarbonylamino-2-(4-benzyloxy-

 phenyl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 10b(i) Preparation of $\mathbf{1 0 b}$ via the Curtius reaction. $\left(4 S, 5 R, 2^{\prime} S\right)$-3-[2-(4-Benzyloxyphenyl)-3-(tert-butoxycarbonyl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 8b (1.21 g, 2.35 mmol ) was converted into the acid $9 b(1.02 \mathrm{~g}, 95 \%)$ by analogy with the preparation of $9 \mathbf{a}$. A portion of the crude acid $9 \mathbf{b}(463 \mathrm{mg}, c a .1$ $\mathbf{m m o l}$ ) was then transformed into the urethane 10b, by analogy with the preparation of $\mathbf{1 0 a}$ using the Curtius reaction. Flash chromatography [EtOAc-petrol $(15: 85)$ to EtOAc-petrol ( $25: 75$ ); gradient elution], afforded the title compound $\mathbf{1 0 b}$ as a white foam ( $241 \mathrm{mg}, 40 \%$ from 8 b ), $\mathrm{mp} 50-51^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}^{35}+90(c 1$, $\left.\mathrm{CHCl}_{3}\right) ; v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 3417,1782,1721$ and $1694 ; \delta_{\mathrm{H}}(250$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $0.92(3 \mathrm{H}, \mathrm{d}, J 7), 3.51-3.83(2 \mathrm{H}, \mathrm{m}), 4.67(1 \mathrm{H}$, quintet, $J 7), 5.01(1 \mathrm{H}, \mathrm{br}$ s), $5.04(2 \mathrm{H}, \mathrm{s}), 5.09(2 \mathrm{H}, \mathrm{s}), 5.16(2$ H, dd, $J 8,6$ ), 5.47 ( $1 \mathrm{H}, \mathrm{d}, J 7$ ), 6.92-6.99 ( $2 \mathrm{H}, \mathrm{m}$ ) and 7.24 $7.47(17 \mathrm{H}, \mathrm{m}) ; m / z\left(\mathrm{FAB}\right.$ ex $\mathrm{Et}_{2} \mathrm{O}$-nitrobenzyl alcohol) 565 $\left(\mathrm{MH}^{+}, 54 \%\right), 521$ (47), 413 (100), 401 (99) and 224 (52).
(ii) Preparation of 10 b via the Mannich reaction with 1[ $N$-(benzyloxycarbonyl)aminomethyl]benzotriazole 4c. A solution of $(4 S, 5 R)$-3-[4-(benzyloxy)phenylacetyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one $\mathbf{6 b}(803 \mathrm{mg}, 2.0 \mathrm{mmol})$ in THF $(10$ ml ) was added by cannula to a 0.35 m solution of $\mathrm{LiN}\left(\mathrm{SiMe}_{3}\right)_{2}$ in THF $(6.2 \mathrm{ml}, 2.2 \mathrm{mmol})$ at $-60^{\circ} \mathrm{C}$. The mixture was treated with a solution of 1-[ $N$-(benzyloxycarbonyl)aminomethyl]benzotriazole $4 \mathbf{c}(565 \mathrm{mg}, 2.0 \mathrm{mmol})$ in THF ( 8 ml ) and was then allowed to warm up to $8^{\circ} \mathrm{C}$ over 3 h . Saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{ml})$ was added, then the mixture was diluted with $\mathrm{Et}_{2} \mathrm{O}(30 \mathrm{ml})$ and washed with $2 \mathrm{~m} \mathrm{HCl}(30 \mathrm{ml})$ followed by aqueous $\mathrm{NaHCO}_{3}(30 \mathrm{ml})$. Drying and evaporation of the organic phase gave a yellow oil ( 1.35 g ). Flash chromatography [ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{Et}_{2} \mathrm{O}$ (96:4); gradient elution] gave two diastereoisomeric Mannich products as follows.

The title compound $\mathbf{1 0 b}(661 \mathrm{mg}, 59 \%)$, mp $51-54^{\circ} \mathrm{C}$, was obtained from the earlier chromatographic fractions and was identical by ${ }^{1} \mathrm{H}$ NMR spectroscopy to a sample of this compound prepared using the Curtius reaction, as described in the preceding experiment. Evaporation of the later fractions from the chromatography gave a colourless oil, which solidified upon standing and was considered to be ( $4 \mathrm{~S}, 5 \mathrm{R}, 2^{\prime} \mathrm{S}$ )-3-[3-benzyloxycarbonylamino-2-(4-benzyloxyphenyl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 11b ( $36 \mathrm{mg}, 3 \%$ ), on the basis of the following properties: $\mathrm{mp} 48-51^{\circ} \mathrm{C}$; $[a]_{\mathrm{D}}^{34}-129.5$ (c $0.93, \mathrm{CHCl}_{3}$ ); $v_{\max }$ (film) $/ \mathrm{cm}^{-1} 3370,1780,1718$ and 1700 ; $\delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 0.76(3 \mathrm{H}, \mathrm{d}, J 7), 3.55-3.80(2 \mathrm{H}, \mathrm{m})$, $4.81(1 \mathrm{H}$, quintet, $J$ 7), $5.00-5.15(6 \mathrm{H}, \mathrm{m}), 5.63(1 \mathrm{H}, \mathrm{d}, J$ 7), 6.95 ( $2 \mathrm{H}, \mathrm{d}, J 8$ ) and 7.17-7.46 ( $17 \mathrm{H}, \mathrm{m}$ ); $m / z$ (FAB ex $\mathrm{Et}_{2} \mathrm{O}$-nitrobenzyl alcohol) $565\left(\mathrm{MH}^{+}, 31 \%\right), 520$ (64), 412 (99) and 400 (100).

## (4S,5R,2'R)-3-[3-Benzyloxycarbonylamino-2-(1-benzyloxycarb-onylindol-3-yl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2one 10c

(i) Preparation of 10c via the Curtius reaction. This was performed by analogy with the preparation of $\mathbf{1 0 b}$ by the Curtius reaction, starting from $\left(4 S, 5 R, 2^{\prime} S\right)$-3-[2-(1-benzyloxycarbon-ylindol-3-yl)-3-(tert-butoxycarbonyl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 8c ( $192 \mathrm{mg}, 0.33 \mathrm{mmol}$ ). Flash chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{EtOAc}(99: 1)\right]$ gave a colourless oil containing a mixture of $\mathbf{1 0 c}$ and benzyl alcohol. Reprecipitation from diethyl ether with petrol afforded the title compound ( $108 \mathrm{mg}, 52 \%$ from 8c) as a white solid, $\mathrm{mp} 63-64^{\circ} \mathrm{C}$ (Found: C, 70.4; H, 5.2; $\mathrm{N}, 6.4 . \mathrm{C}_{37} \mathrm{H}_{33} \mathrm{~N}_{3} \mathrm{O}_{7}$ requires $\mathrm{C}, 70.35 ; \mathrm{H}, 5.3$; N, $6.65 \%) ;[a]_{\mathrm{D}}^{30}+74.7\left(\right.$ c $\left.0.79, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 3400$, $1784,1731,1696$ and $1678 ; \delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 0.93(3 \mathrm{H}$, d, $J$ 7), 3.68-3.96 ( $2 \mathrm{H}, \mathrm{m}$ ), $4.70(1 \mathrm{H}$, quintet, $J$ 7), $5.05-5.12$ $(3 \mathrm{H}, \mathrm{m}), 5.39-5.52(4 \mathrm{H}, \mathrm{m}), 7.25-7.50(17 \mathrm{H}, \mathrm{m}), 7.67(1 \mathrm{H}$, s), $7.76(1 \mathrm{H}, \mathrm{d}, J 8)$ and $8.18(1 \mathrm{H}, \mathrm{d}, J 8)$; $m / z(\mathrm{FAB} e x$ $\mathrm{Et}_{2} \mathrm{O}$-nitrobenzyl alcohol) $632\left(\mathrm{MH}^{+}, 100 \%\right), 588$ (85), 467 (39) and 306 (8).
(ii) Preparation of 10c via the Mannich reaction with 1-[ $N$ (benzyloxycarbonyl)aminomethyl]benzotriazole 4c. This was performed by analogy with the preparation of $\mathbf{1 0 b}$ by method (ii), starting from ( $4 S, 5 R$ )-3-[(1-benzyloxycarbonylindol-3-yl)acetyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one $\mathbf{6 c}(627 \mathrm{mg}$, 1.34 mmol ). Evaporation of the crude product in vacuo left a pale yellow foam ( 894 mg ). A portion of this residue ( 13 mg ) was analysed by $250 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy, on the basis of which compounds $\mathbf{1 0 c}, 5, \mathbf{6 c}$ and 11 c were judged to be present in the approximate molar ratio $80: 10: 6: 4$. The remainder of the residue from the evaporation was purified by flash chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ to $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ - EtOAc (96:4); gradient elution] to yield the following three fractions. First fraction: ( $4 S, 5 R$ )-3-[(1-benzyloxycarbonylindol-3-yl)acetyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 6c ( $37.5 \mathrm{mg}, 6 \%$ ), identical to the starting material by ${ }^{1} \mathrm{H}$ NMR spectroscopy. Second fraction: $\left(4 S, 5 R, 2^{\prime} R\right)$-3-[3-(Benzyloxycarbonylamino)-2-(1-benzyloxycarbonylindol-3-yl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 10c as a pale yellow foam ( 484 mg , $58 \%$ ), mp $59-65^{\circ} \mathrm{C}$, identical by ${ }^{1} \mathrm{H}$ NMR spectroscopy ( 250 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) to material prepared by the Curtius reaction, as described in the preceding experiment. Third fraction: considered to be (4S,5R,2'S)-3-[3-benzyloxycarbonylamino-2-(1-benzyloxycarbonylindol-3-yl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 11c ( $36 \mathrm{mg}, 3 \%$ ), on the basis of the following data: off-white foam, $\mathrm{mp} 63-66^{\circ} \mathrm{C}$; $[a]_{\mathrm{D}}^{34}-95$ (c 0.73 , $\mathrm{CHCl}_{3}$ ); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3378,1780,1726$ and 1698 (shoulder); $\delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 0.76(3 \mathrm{H}, \mathrm{d}, J 7), 3.65-3.90(2 \mathrm{H}$, m ), $4.80(1 \mathrm{H}$, quintet, $J 7$ ), $5.06-5.18(3 \mathrm{H}, \mathrm{m}), 5.39-5.50(3$ $\mathrm{H}, \mathrm{m}), 5.66(1 \mathrm{H}, \mathrm{d}, J 7), 7.16-7.52(17 \mathrm{H}, \mathrm{m}), 7.60(1 \mathrm{H}, \mathrm{s})$, $7.78(1 \mathrm{H}, \mathrm{d}, J 8)$ and $8.18(1 \mathrm{H}, \mathrm{d}, J 8) ; m / z\left(\mathrm{FAB}\right.$ ex $\mathrm{Et}_{2} \mathrm{O}-$ nitrobenzyl alcohol) $632\left(\mathrm{MH}^{+}, 47 \%\right), 588$ (51), 479 (82) and 467 (100).
(4S,5R,2'R)-3-(3-Benzyloxycarbonylamino-2-benzylpropanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 10d
(i) Preparation of $10 \mathrm{~d} v i a$ the Curtius reaction. This was performed by analogy with the preparation of 10a by method (i), starting from ( $4 S, 5 R, 2^{\prime} R$ )-3-(2-benzyl-3-carboxypropanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 9d ( $336 \mathrm{mg}, 0.9 \mathrm{mmol}$ ). The crude product was purified by flash chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{Et}_{2} \mathrm{O}(19: 1)\right]$ to give a colourless oil which was recrystallised from diethyl ether and petrol to yield the title compound 10 d as white crystals ( $252 \mathrm{mg}, 60 \%$ ), mp $91-93^{\circ} \mathrm{C}$; $[a]_{\mathrm{D}}^{31}+11.0\left(c 1, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 3377,1778,1737$ and $1690 ; \delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 0.83(3 \mathrm{H}, \mathrm{d}), 2.86(1 \mathrm{H}$, dd, $J 14,7$ ), 3.02 ( $1 \mathrm{H}, \mathrm{dd}, J 13,8$ ), $3.43-3.64(2 \mathrm{H}, \mathrm{m}), 4.39(1 \mathrm{H}, \mathrm{br}$ quintet, $J$ 7), $4.54(1 \mathrm{H}$, quintet, $J$ 7), $5.09(2 \mathrm{H}, \mathrm{s}), 5.12-5.22$ ( 1 $\mathrm{H}, \mathrm{br}), 5.27(1 \mathrm{H}, \mathrm{d}, J 7)$ and $7.20-7.45(15 \mathrm{H}, \mathrm{m}) ; m / z 472\left(\mathrm{M}^{+}\right.$, $4 \%$ ), 381 (6), 335 (10), 308 (59), 160 (30), 131 (46) and 91 (100) (Found: $\mathrm{M}^{+}, 472.1997 . \mathrm{C}_{28} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{5}$ requires $M, 472.1998$ ).
(ii) Preparation of $10 \mathrm{~d} v i a$ the Mannich reaction with benzyl N -(acetoxymethyl)carbamate. This was performed analogously to the preparation of 10a by method (iii), starting from $\mathbf{8 d}$ (691 $\mathrm{mg}, 2.23 \mathrm{mmol}$ ). The reaction mixture was allowed to warm up from -78 to $0^{\circ} \mathrm{C}$ over 3 h , then was kept at $0^{\circ} \mathrm{C}$ for 3 h and at $20^{\circ} \mathrm{C}$ for 1 h after which work up was performed as before. Analysis of the crude product by $250 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy indicated that the starting material 8d and the Mannich product 10d were present in a $c a .1: 1$ molar ratio. Flash chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{Et}_{2} \mathrm{O}\right.$ (97:3)] gave first 8d ( $292 \mathrm{mg}, 42 \%$ recovery) followed by $\mathbf{1 0 d}(446 \mathrm{mg}$ ) as an oil. Recrystallisation of the latter compound from diethyl etherpetrol gave 10d ( $400 \mathrm{mg}, 38 \% ; 66 \%$ based on recovered $\mathbf{8 d}$ ) as white crystals, mp $90-93{ }^{\circ} \mathrm{C}$, identical by $250 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy with material prepared by the Curtius reaction according to the preceding experiment.
(4S,5R,2'R)-3-(3-Benzyloxycarbonylamino-2-methylpropanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 10e via the Mannich reaction with benzyl N -(acetoxymethyl)carbamate
This was prepared analogously to the preparation of $\mathbf{1 0 a}$ by method (iii), starting from (4S,5R)-4-methyl-5-phenyl-3-propanoyl-1,3-oxazolidin-2-one $6 \mathrm{e} .{ }^{25}$ The reaction mixture was allowed to warm from -78 to $20^{\circ} \mathrm{C}$ over 8 h and then was stirred at $20^{\circ} \mathrm{C}$ for 1 h , before being worked up as before. Flash chromatography [EtOAc-petrol; gradient from $(2: 1)$ to $(1: 1)$ ] yielded (4S,5R,2'R)-3-(3-benzyloxycarbonylamino-2-methyl-propanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one $\mathbf{1 0 e}(130 \mathrm{mg}$, $34 \%$ ) as a clear glass. Recrystallisation from petrol gave white crystals, mp 91-93 ${ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}^{29}-33.7\left(c 2.1, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ; \delta_{\mathrm{H}}(250 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 0.86(3 \mathrm{H}, \mathrm{d}, J 7), 1.23(3 \mathrm{H}, \mathrm{d}, J 7), 3.41-3.54(2 \mathrm{H}, \mathrm{m})$, $3.96(1 \mathrm{H}$, sextet, $J 7), 4.73(1 \mathrm{H}$, quintet, $J 7), 5.08(2 \mathrm{H}, \mathrm{s})$, 5.19-5.29 ( 1 H , br t), $5.65(1 \mathrm{H}, \mathrm{d}, J 7)$ and $7.25-7.45(10 \mathrm{H}, \mathrm{m})$; $m / z 369\left(\mathrm{M}^{+}, 9 \%\right), 289$ (5), 219 (8) and 91 (100) (Found: $\mathbf{M}^{+}$, 396.1692. $\mathrm{C}_{22} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{5}$ requires $M, 396.1685$ ).

## ( $R$ )-3-Benzyloxycarbonylamino-2-phenylpropanoic acid 12a

 ( $4 S, 5 R, 2^{\prime} R$ )-3-(3-Benzyloxycarbonylamino-2-phenylpropan-oyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 10a (212 mg, 0.46 mmol ) was dissolved in a mixture of THF $(4 \mathrm{ml})$ and $\mathrm{H}_{2} \mathrm{O}$ $(1.3 \mathrm{ml})$ and cooled to $0^{\circ} \mathrm{C} . \mathrm{LiOH} \cdot \mathrm{H}_{2} \mathrm{O}(49 \mathrm{mg}, 1.16 \mathrm{mmol})$, and aqueous $\mathrm{H}_{2} \mathrm{O}_{2}(30 \%$ solution, 0.25 ml$)$ were added. The reaction was quenched after 30 min with saturated aqueous $\mathrm{Na}_{2} \mathrm{SO}_{3}(10 \mathrm{ml})$ and diluted with brine $(10 \mathrm{ml})$.The reaction mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 15 \mathrm{ml})$. The aqueous layer was then acidified with 2 m HCl (to pH 2 ) and extracted into EtOAc ( $3 \times 15 \mathrm{ml}$ ). The combined organic layers were dried and evaporated under vacuum to yield a viscous oil, which was recrystallised from diethyl ether and petrol to yield the title compound as white crystals ( $89.4 \mathrm{mg}, 65 \%$ ), mp $95-97{ }^{\circ} \mathrm{C}$ (lit., ${ }^{26} 100-102^{\circ} \mathrm{C}$ ); $[a]_{\mathrm{D}}^{36}+92.5$ (c 1.0 in EtOH) $\left[\right.$ lit. ${ }^{26}{ }^{26}+93.6$ (c 1.0, EtOH)]; $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 3390,1720$ and $1654 ; \delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 3.65-4.15(3 \mathrm{H}, \mathrm{m}), 5.15-5.35(3 \mathrm{H}$,
$\mathrm{m})$ and $7.35-7.55(10 \mathrm{H}, \mathrm{m})$. The broadness of the peaks in the ${ }^{1} \mathrm{H}$ NMR spectrum was attributed to the presence of interconverting rotamers.

## ( $R$ )-3-Benzyloxycarbonylamino-2-(4-benzyloxyphenyl)propanoic acid 12b

This was prepared by analogy with 12a, starting from (4S,5R, 2' $R$ )-3-[3-benzyloxycarbonylamino-2-(4-benzyloxy-phenyl)propanoyl]-4-methyl-5-phenyl-1,3-oxazolidin-2-one 10b $(571 \mathrm{mg}, 1.01 \mathrm{mmol})$. Recrystallisation from $\mathrm{Et}_{2} \mathrm{O}$-petrol yielded the title compound $\mathbf{1 2 b}$ as a white solid $(283 \mathrm{mg}$, $69 \%$ ), mp $105-108{ }^{\circ} \mathrm{C}$ (from $\mathrm{Et}_{2} \mathrm{O}$-petrol) (Found: C, 71.0; H, 5.6; N , 3.3. $\mathrm{C}_{24} \mathrm{H}_{23} \mathrm{NO}_{5}$ requires $\mathrm{C}, 71.1 ; \mathrm{H}, 5.7 ; \mathrm{N}, 3.45 \%$ ); $[a]_{\mathrm{D}}^{28}+113\left(c 1.1, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 3350,2400-3500$ (br) and $1697 ; \delta_{\mathrm{H}}\left(250 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$, broad spectrum, probably as a result of the presence of interconverting rotamers) 3.5-3.9 (3 H, m), $5.05(2 \mathrm{H}, \mathrm{s}), 5.05-5.20(2 \mathrm{H}, \mathrm{m}), 6.05(1 \mathrm{H}$, br s, exchanges with $\left.\mathrm{D}_{2} \mathrm{O}\right), 6.93(2 \mathrm{H}, \mathrm{d}, J 8), 7.13-7.22(2 \mathrm{H}$, $\mathrm{m})$ and $7.30-7.45(10 \mathrm{H}, \mathrm{m}) ; m / z 405\left(\mathrm{M}^{+}, 0.1 \%\right), 254$ (11) and 91 (100) (Found: $\mathrm{M}^{+}, 405.1576 . \mathrm{C}_{24} \mathrm{H}_{23} \mathrm{NO}_{5}$ requires $M$, 405.1564).

## (R)-3-Benzyloxycarbonylamino-2-(1-benzyloxycarbonylindol-3yl)propanoic acid 12c

(4S,5R, 2' $R$ )-3-[3-Benzyloxycarbonylamino-2-(1-benzyloxy-carbonylindol-3-yl)propanoyl]-4-methyl-5-phenyl-1,3-oxazol-idin-2-one $10 \mathrm{c}(50.2 \mathrm{mg}, 0.079 \mathrm{mmol})$ was dissolved in THF (1 $\mathrm{ml})$ and the solution was cooled to $0{ }^{\circ} \mathrm{C}$; $\mathrm{LiOH} \cdot \mathrm{H}_{2} \mathrm{O}(4.0 \mathrm{mg}$, $0.095 \mathrm{mmol})$, aqueous $\mathrm{H}_{2} \mathrm{O}_{2}(30 \%, 0.1 \mathrm{ml})$ and water $(0.3 \mathrm{ml})$ were then added. After 25 min aqueous $\mathrm{Na}_{2} \mathrm{SO}_{3}(1 \mathrm{ml})$ was added followed by 2 m HCl to pH 3 . The mixture was partitioned between EtOAc ( 10 ml ) and water $(10 \mathrm{ml})$. Drying and evaporation of the EtOAc phase gave an off-white solid which was recrystallised from $\mathrm{CH}_{2} \mathrm{Cl}_{2}$-petrol to give the title compound 12c ( $31.5 \mathrm{mg}, 84 \%$ ), $\mathrm{mp} 163-164^{\circ} \mathrm{C}$ (Found: C, $68.5 ; \mathrm{H}$, $5.1 ; \mathrm{N}, 5.9 . \mathrm{C}_{27} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{6}$ requires C, $\left.68.6 ; \mathrm{H}, 5.1 ; \mathrm{N}, 5.9 \%\right) ;[a]_{\mathrm{D}}^{35}$ +58 (c 1.0, acetone); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 3331,2500-3300$ (br), 1743,1719 and $1684 ; \delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 3.5-3.8(2 \mathrm{H}, \mathrm{m})$, 4.0-4.3 (1 H, m), 5.0-5.3 (3 H, m), 5.4-5.5 (2 H, m), 7.05-7.7 $(14 \mathrm{H}, \mathrm{m})$ and $8.17(1 \mathrm{H}, \mathrm{d}, J 8) ; \mathrm{m} / \mathrm{z}$ (FAB ex petrol-glycerol) $473\left(\mathrm{MH}^{+}, 16 \%\right), 429$ (17), 283 (17) and 214 (100).

## ( $R$ )-3-Benzyloxycarbonylamino-2-benzylpropanoic acid 12d

This was prepared by analogy with $\mathbf{1 2 a}$, from $\left(4 S, 5 R, 2^{\prime} R\right)$-3-(3-benzyloxycarbonylamino-2-benzylpropanoyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one $\mathbf{1 0 d}(225 \mathrm{mg}, 0.5 \mathrm{mmol})$. The crude product was crystallised from $\mathrm{Et}_{2} \mathrm{O}$-petrol to yield the title compound 12d $(99.6 \mathrm{mg}, 67 \%)$ as white crystals. After a further recrystallisation from $\mathrm{Et}_{2} \mathrm{O}$-petrol the crystals $(61 \mathrm{mg})$ had mp $74-75^{\circ} \mathrm{C}$ (Found: C, $68.8 ; \mathrm{H}, 6.0 ; \mathrm{N}, 4.4 . \mathrm{C}_{18} \mathrm{H}_{19} \mathrm{NO}_{4}$ requires C, $69.0 ; \mathrm{H}, 6.1 ; \mathrm{N}, 4.5 \%) ;[\alpha]_{\mathrm{D}}^{27}+9.7\left(c 0.15, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;[a]_{\mathrm{D}}^{27}-1.4(c$ $0.4, \mathrm{EtOH}) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 3348,2400-3600(\mathrm{br}), 1792$ and 1695; $\delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.62-3.55(5 \mathrm{H}, \mathrm{m}), 5.02-5.22(3 \mathrm{H}$, $\mathrm{m})$ and $7.10-7.40(10 \mathrm{H}, \mathrm{m})$; the peaks in the NMR spectrum were broad, probably as a consequence of the presence of interconverting rotamers; $m / z 313\left(\mathrm{M}^{+}, 0.3 \%\right)$, 222 (0.4), 131 (7), 91 (100) and 65 (21) (Found: $\mathrm{M}^{+}, 313.1317 . \mathrm{C}_{18} \mathrm{H}_{19} \mathrm{NO}_{4}$ requires M, 313.1314).

## ( $R$ )-3-Amino-2-phenylpropanoic acid 13

( $4 S, 5 R, 2^{\prime} R$ )-3-(3-Benzyloxycarbonylamino-2-phenylpropan-oyl)-4-methyl-5-phenyl-1,3-oxazolidin-2-one 10 a ( $459 \mathrm{mg}, 1$ $\mathrm{mmol})$ was hydrolysed using $\mathrm{LiOH} \cdot \mathrm{H}_{2} \mathrm{O}(104 \mathrm{mg}, 2.5 \mathrm{mmol})$ as in the preparation of $\mathbf{1 2 a}$. The crude ( $R$ )-3-benzyloxycarbonyl-amino-2-phenylpropanoic acid 12a was not recrystallised, but was dissolved in glacial acetic acid $(10 \mathrm{ml}) .10 \%$ Palladium on carbon $(300 \mathrm{mg})$ was added and the mixture was hydrogenated overnight. Filtration through Celite, repeated evaporation with water and recrystallisation from water-ethanol gave the title compound 13 as white crystals ( $112 \mathrm{mg}, 68 \%$ from 10a), mp
$222-224{ }^{\circ} \mathrm{C}$ (lit., ${ }^{20} 224-225{ }^{\circ} \mathrm{C}$, lit., ${ }^{26} 223-226^{\circ} \mathrm{C}$ ); $[a]_{\mathrm{D}}^{30}+94(c$ $\left.0.2, \mathrm{H}_{2} \mathrm{O}\right)\left[\right.$ lit. $.{ }^{20}+95\left(c 0.2\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}\right)$, lit., ${ }^{26}+85\left(c 0.2\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}\right)$ ]; $\delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) 3.30(1 \mathrm{H}, \mathrm{dd}, J 12,7), 3.49(1 \mathrm{H}, \mathrm{dd}, J 12$, $8), 3.81(1 \mathrm{H}, ~ ' \mathrm{t}$ ', J 7) and $7.34-7.48(5 \mathrm{H}, \mathrm{m})$.

## ( $R$ )-3-Amino-2-phenylpropanoic acid, (1S)-(+)-camphorsulfonate salt

A solution of $(R)$-3-amino-2-phenylpropanoic acid 13 (32.0 $\mathrm{mg}, 0.19 \mathrm{mmol}$ ) in water $(4 \mathrm{ml})$ was treated with $(1 S)-(+)-$ camphor-10-sulfonic acid ( $49.0 \mathrm{mg}, 0.21 \mathrm{mmol}$ ) in $\mathrm{EtOH}(1 \mathrm{ml})$. The solvents were evaporated and the residue was twice recrystallised from $\mathrm{EtOH}-\mathrm{Et}_{2} \mathrm{O}$ to give the title compound ( 52.5 $\mathrm{mg}, 68 \%$ ) as white needles, $\mathrm{mp} 183-189{ }^{\circ} \mathrm{C}$ (lit., ${ }^{26} 190-192{ }^{\circ} \mathrm{C}$ ); $[a]_{\mathrm{D}}^{26}+61\left(c 0.5\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}\right)\left[\right.$ lit., $\left.{ }^{26}+63\left(c 0.5, \mathrm{H}_{2} \mathrm{O}\right)\right]$. The relative configuration of the title compound was proven by single crystal X-ray diffraction. ${ }^{6}$

## (R)-3-Amino-2-(4-hydroxyphenyl)propanoic acid 14

(R)-3-(Benzyloxycarbonylamino)-2-(4-benzyloxyphenyl)propanoic acid 12b $(107.7 \mathrm{mg})$ was dissolved in glacial acetic acid (2 $\mathrm{ml})$ and hydrogenated over Pd black ( 15 mg ) for 1 h , during which time white crystals formed in the reaction mixture. Water $(2 \mathrm{ml})$ was added and the mixture was filtered through Celite, which was then washed with $50 \%$ aqueous $\mathrm{AcOH}(10 \mathrm{ml})$. Evaporation of the combined filtrate and washings, then crystallisation from water, yielded the title compound $14(43.7 \mathrm{mg}$, $91 \%$ ) as a white solid, mp ca. $270^{\circ} \mathrm{C}$ (decomp.) (Found: C, 59.2; $\mathrm{H}, 6.0 ; \mathrm{N}, 7.5 . \mathrm{C}_{9} \mathrm{H}_{11} \mathrm{NO}_{3}$ requires $\left.\mathrm{C}, 59.7 ; \mathrm{H}, 6.2 ; \mathrm{N}, 7.6 \%\right) ;[\alpha]_{\mathrm{D}}^{30}$ +16.7 ( c 1.2, $\left.\mathrm{HCO}_{2} \mathrm{H}\right)$; $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 3225,2400-3300(\mathrm{br})$, 1653,1589 and $1510 ; \delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{D}\right) 3.62-3.76(1 \mathrm{H}$, $\mathrm{m}), 3.84-3.95(1 \mathrm{H}, \mathrm{m}), 4.33(1 \mathrm{H}, \mathrm{t}, J 7), 6.8(2 \mathrm{H}, \mathrm{d}, J 8), 7.1(1$ $\mathrm{H}, \mathrm{br} \mathrm{s})$ and $7.32(2 \mathrm{H}, \mathrm{d}, J 8) ; m / z 181\left(\mathrm{M}^{+}, 2 \%\right), 164$ (15), 152 (100) and 107 (64) (Found: $\mathrm{M}^{+}$, 181.0737. $\mathrm{C}_{9} \mathrm{H}_{11} \mathrm{NO}_{3}$ requires $M, 181.0739$ ).

## (R)-3-Amino-2-benzylpropanoic acid 15

This was prepared by analogy with $\mathbf{1 3}$, starting from $(R)$-3-benzyloxycarbonylamino-2-benzylpropanoic acid 12d (53.2 $\mathrm{mg}, 0.17 \mathrm{mmol}$ ). Crystallisation from $\mathrm{H}_{2} \mathrm{O}-\mathrm{EtOH}$ yielded the title compound 15 ( $24.6 \mathrm{mg}, 81 \%$ ) as white crystals, mp 227$229{ }^{\circ} \mathrm{C}$ (lit., ${ }^{2} 225-226^{\circ} \mathrm{C}$ ); $[\alpha]_{\mathrm{D}}^{35}+18.9(c 0.88,1 \mathrm{~m} \mathrm{HCl}),[\alpha]_{\mathrm{D}}^{33}$ $+19.2(c 0.65,1 \mathrm{~m} \mathrm{HCl})\left\{\right.$ lit., $\left.{ }^{2}[\alpha]_{\mathrm{D}}^{29}+11.3(c 1,1 \mathrm{~m} \mathrm{HCl})\right\} ; \delta_{\mathrm{H}}(250$ $\left.\mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}\right) 2.90-3.27(5 \mathrm{H}, \mathrm{m})$ and 7.41-7.57 $(5 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}(63$ $\mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}$ ) 39.2, 43.8, 50.1, 129.7, 131.8, 132.0, 141.7 and 182.7; m/z $179\left(\mathrm{M}^{+}, 53 \%\right), 162$ (51), 144 (9), 117 (75), 103 (22), 91 (100), 78 (30) and 65 (29) (Found: $\mathrm{M}^{+}$, 179.0947. $\mathrm{C}_{10} \mathrm{H}_{13} \mathrm{NO}_{2}$ requires $M, 179.0946)$.

## (1-Benzyloxycarbonylindol-3-yl)acetic acid 18

Indole-3-acetic acid $17(4.20 \mathrm{~g}, 24 \mathrm{mmol})$ was dissolved in dry THF ( 100 ml ) and the solution was cooled to $-45^{\circ} \mathrm{C}$. A 2.2 m solution of BuLi in hexane ( $24 \mathrm{ml}, 53 \mathrm{mmol}$ ) was added over 5 min . After a further 5 min benzyl chloroformate ( $3.6 \mathrm{ml}, 24$ mmol ) was added and the reaction mixture was allowed to attain $-5^{\circ} \mathrm{C}$ over 2 h before being quenched with saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}(10 \mathrm{ml})$. Water $(100 \mathrm{ml})$ was added and the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(2 \times 60 \mathrm{ml})$. The aqueous phase was acidified with 2 m HCl to pH 3 and was then extracted with diethyl ether ( $3 \times 60 \mathrm{ml}$ ). The ether extracts from after the acidification were combined, diluted with sufficient EtOAc to dissolve any precipitate, dried, filtered and evaporated. Recrystallisation from EtOAc-petrol, together with flash chromatography $\left[\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{EtOAc}(4: 1)\right.$ to $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{EtOAc}$ ( $2: 1$ ); gradient elution] of the residue from evaporation of the mother liquors, and further recrystallisation $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}-\right.$ petrol $)$ yielded the title compound $\mathbf{1 8}(5.14 \mathrm{~g}, 69 \%)$ as white crystals, mp 153-154 ${ }^{\circ} \mathrm{C}$ (Found: C, 69.8; H, 4.9; N, 4.55. $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{NO}_{4}$ requires $\mathrm{C}, 69.9 ; \mathrm{H}, 4.9 ; \mathrm{N}, 4.5 \%) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 2400-3600$ (br), 1733 and $1696 ; \delta_{\mathrm{H}}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 3.75(2 \mathrm{H}, \mathrm{s}), 5.46$ $(2 \mathrm{H}, \mathrm{s}), 7.23-7.55(8 \mathrm{H}, \mathrm{m})$ and $8.17(1 \mathrm{H}, \mathrm{d}, J 8) ; m / z 309$
$\left(\mathrm{M}^{+}, 18 \%\right), 265$ (5), 220 (18) and 91 (100) (Found: $\mathrm{M}^{+}$, 309.1000. $\mathrm{C}_{18} \mathrm{H}_{15} \mathrm{NO}_{4}$ requires $M, 309.1001$ ).

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